

N[anoscale therm](http://rileyhanus.com/science.html)al transport

Lecture 3

Riley Hanus http://rileyhanus.com/science.html

- 1. General discussion about heat capacity
- 2. Obtain the phonon heat capacity
	- a) examine limiting behavior
	- b) low ω , high T, low T
- 3. Heat capacity in practice
	- a) Debye level
	- b) Debye temperature
	- c) Speed sound

Heat capacity \vert Heat capacity is generally a very boring material property.

 $n:$ number density of atoms

 y : Gruneisen parameter

 $\alpha_{\rm V}$: volumetric thermal expansion ($\alpha_{\rm V} \cong 3\alpha_{\rm L}$) $4\overline{a}$

on board

Derivation set-up

Heat capacity is the change in energy with temperature

We will look for the *spectral* heat capacity *per unit volume*

$$
C(\omega) = \frac{d E(\omega, T)}{dT}
$$

Energy density: $E(\omega, T) = \hbar \omega n(\omega, T)$

Number density of phonons: $n(\omega, T) = g(\omega) n_{\text{BE}}(\omega, T)$

Note:

- $g(\omega)$ is the number of possible states per unit volume. Here, we say it doesn't change with temperature (i.e. the quasi-harmonic approximation). In reality thinks get softer with increasing temperature so $q(\omega)$ will shift down in ω as T increases.
- $n_{\text{BE}}(\omega, T)$ is the occupation number (can think of it as an occupation probability), of a state at frequency/energy ω , and temperature T.

$$
\frac{d E(\omega, T)}{dT} = \hbar \omega g(\omega) \frac{d n_{BE}(\omega, T)}{dT}
$$

on board

Derivation

$$
n_{\text{BE}} = \frac{1}{e^{\frac{\hbar \omega}{k_{\text{B}}T}} - 1}
$$

Crazy triple chain rule

$$
\frac{dn_{BE}}{dT} = \frac{(-1)}{\left(\frac{\hbar\omega}{e^{k_BT}-1}\right)^2} \times e^{\frac{\hbar\omega}{k_BT}} \times \frac{(-1)\hbar\omega}{k_BT^2} = \frac{\hbar\omega}{k_BT^2} \frac{e^{\frac{\hbar\omega}{k_BT}}}{\left(e^{\frac{\hbar\omega}{k_BT}-1}\right)^2}
$$

$$
C(\omega) = \hbar \omega g(\omega) \frac{dn_{BE}}{dT} = \hbar \omega \times \frac{3}{2\pi^2} \frac{\omega^2}{v_p^2 v_g} \times \frac{\hbar \omega}{k_B T^2} \frac{\frac{\hbar \omega}{e^k T}}{\left(e^{\frac{\hbar \omega}{k_B T}} - 1\right)^2}
$$

$$
C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\text{g}} v_{\text{p}}^2} \frac{\hbar^2}{k_{\text{B}} T^2} \frac{e^{\hbar \omega / k_{\text{B}} T}}{(e^{\hbar \omega / k_{\text{B}} T} - 1)^2}
$$

Heat capacity per volume
\n
$$
C = \int_{0}^{\omega_{\text{max}}} C(\omega) d\omega
$$

Limiting behavior \vert Let's look at the low ω (and high T) behavior

$$
C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\text{g}} v_{\text{p}}^2} \frac{\hbar^2}{k_{\text{B}} T^2} \frac{e^{\hbar \omega / k_{\text{B}} T}}{(e^{\hbar \omega / k_{\text{B}} T} - 1)^2}
$$

If I just plug in $\omega = 0$, $C(\omega) = 0/0$. Not super helpful, what we really want is the limiting behavior.

Define:
$$
x = \frac{\hbar \omega}{k_B T}
$$
 Taylor expand e^x about $x = 0$.
\n
$$
C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_p^2 v_g} \frac{\hbar^2}{k_B T^2} \frac{(1)}{(1 + x - 1)^2} = \frac{3}{2\pi^2} \frac{\omega^4}{v_p^2 v_g} \frac{\hbar^2}{k_B T^2} \left(\frac{k_B T}{\hbar \omega}\right)^2 = k_B \frac{3}{2\pi^2} \frac{\omega^2}{v_p^2 v_g}
$$
\nAt high T and or low ω
\n
$$
C(\omega) = k_B g(\omega)
$$
\nwhich means $C \to \text{cst at high T}$

and $C(\omega) \propto \omega^2$ when the dispersion is linear

Limiting behavior $|$ Let's look at **low** T behavior

Convenient to convert the integral to be over the dimensionless parameter x .

$$
x = \frac{\hbar \omega}{k_{\text{B}}T}
$$

\n
$$
c = \frac{3}{2\pi^2} \frac{\hbar^2}{k_{\text{B}}T^2} \int_0^{\omega_{\text{max}}} \frac{\omega^4}{v_{\text{g}}v_{\text{p}}^2} \frac{e^{\hbar \omega / k_{\text{B}}T}}{(e^{\hbar \omega / k_{\text{B}}T} - 1)^2} d\omega
$$

\n
$$
\omega = \frac{k_{\text{B}}T}{\hbar} \frac{dx}{k_{\text{B}}T}
$$

\n
$$
c = \frac{3}{2\pi^2} \frac{k_{\text{B}}^4 T^3}{\hbar^3} \int_0^{x_{\text{max}}} \frac{x^4}{v_{\text{g}}v_{\text{p}}^2} \frac{e^x}{(e^x - 1)^2} dx
$$

At low T:

1. only low frequency modes are populated, $v_{\rm g} = v_{\rm p} = v_{\rm s}$

2. $x_{\text{max}} \rightarrow \infty$

$$
\int_{0}^{\infty} \frac{x^4 e^x}{(e^x - 1)^2} dx = \frac{4\pi^4}{15} \qquad C = \frac{3}{2\pi^2} \frac{k_B^4 T^3}{\hbar^3} \left(\frac{4\pi^4}{v_s^3 15}\right)
$$

At low T where dispersion is linear: $C \propto T^3$

The isotropic average speed of sound, used for thermal conductivity modeling, heat capacity modeling, Debye temperature.

- Speed of sound | when the cross-correlation is used to measure the time delay, the error is typically limit by the thickness variation of the sample. Can get better than 1% accuracy, if careful.
	- If sample is a randomly oriented polycrystalline, then you are measuring the isotropic average longitudinal and transverse speeds of sound.
	- Be mindful of preferred orientation of grains, especially for non-cubic materials.
	- Be mindful of sample density. Samples should be very dense, >98%.

$$
v_{\rm S} = \left(\frac{1}{3} \left[\frac{1}{v_{\rm L}^3} + \frac{2}{v_{\rm T}^3} \right] \right)^{-1/3}
$$

Anderson, O. L. A simplified method for calculating the Debye temperature from elastic constants. *J. Phys. Chem. Solids* **24**, 909–917 (1963). 11

Thermal expansion contribution

 $n:$ number density of atoms $\gamma = B\alpha_{V}/C_{V}$: Gruneisen parameter : isothermal Bulk modulus $\alpha_{\rm V}$: volumetric thermal expansion ($\alpha_{\rm V} \cong 3\alpha_{\rm L}$)

Analyzing high-T data

- Absolute value of high temperature heat capacity is notoriously hard to measure.
- In contrast low-temperature heat capacity can be very accurate (e.g. physical property [measurement system PPMS](https://www.sciencedirect.com/science/article/pii/S2542529318301147) in heat capacity mode).
- Often times our models are better at predicting it then and one given measurement is.
- The slope of Heat Capacity vs. T at high T can be reasonably accurate.

Thermal conductivity considerations

Modeling:

$$
\kappa = \int C(\omega) D(\omega) d\omega
$$

$$
D(\omega) = \frac{1}{3} v_{\rm g}^2(\omega) \tau(\omega)
$$

 $C(\omega)$: spectral heat capacity per unity volume $D(\omega)$: spectral diffusivity

Experiment:

$$
\kappa = C_{\rm V} D = C_{\rm P} \rho D
$$

 D : thermal diffusivity measured by, for example LFA C_V : heat capacity per unit volume $C_{\rm P}$: heat capacity per unit mass ρ : mass density

$$
C_V = \int C(\omega) d\omega
$$
 Heat capacity weighted average diffusivity

$$
D = \frac{\int C(\omega)D(\omega)d\omega}{\int C(\omega)d\omega} \qquad \text{Heat capacity weighted average diffusivity}
$$

$$
\kappa = \left(\int C(\omega) d\omega\right) \frac{\int C(\omega)D(\omega)d\omega}{\int \int C(\omega)d\omega} = \int C(\omega)D(\omega)d\omega
$$

 $C = \int C(x) dx$