

# Nanoscale thermal transport

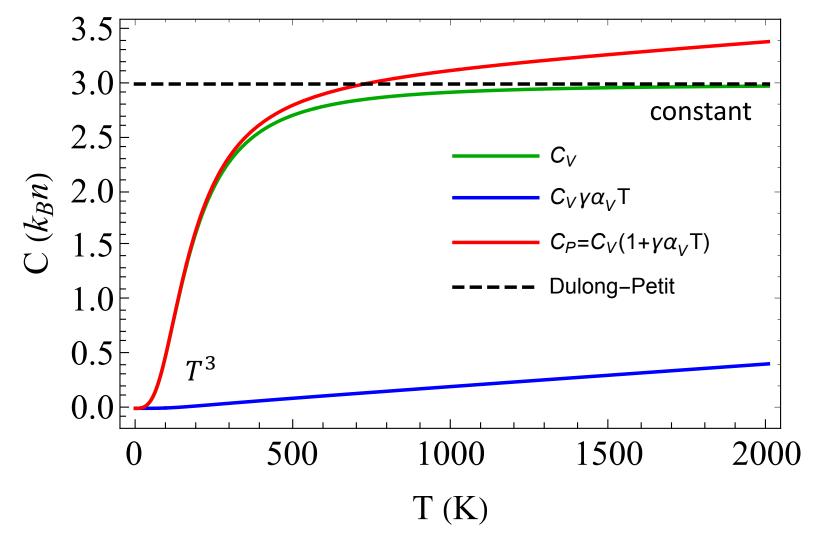
Lecture 3

**Riley Hanus** 

http://rileyhanus.com/science.html

- 1. General discussion about heat capacity
- 2. Obtain the phonon heat capacity
  - a) examine limiting behavior
  - b) low  $\omega$ , high T, low T
- 3. Heat capacity in practice
  - a) Debye level
  - b) Debye temperature
  - c) Speed sound

Material	Volumetric heat capacity at 25 C $C_V (J/K \text{ cm}^3)$
Liquid Hg	1.9
Liquid water	4.2
Cu	3.45
Glass	2.1
Al	2.4
Graphite	1.53
Diamond	1.8



*n*: number density of atoms

 $\gamma$ : Gruneisen parameter

 $\alpha_{\rm V}$ : volumetric thermal expansion ( $\alpha_{\rm V} \cong 3\alpha_{\rm L}$ )

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#### **Derivation set-up**

Heat capacity is the change in energy with temperature

We will look for the *spectral* heat capacity *per unit volume* 

$$C(\omega) = \frac{d E(\omega, T)}{dT}$$

Energy density:  $E(\omega, T) = \hbar \omega n(\omega, T)$ 

Number density of phonons:  $n(\omega, T) = g(\omega) n_{BE}(\omega, T)$ 

Note:

- g(ω) is the number of possible states per unit volume. Here, we say it doesn't change with temperature (i.e. the quasi-harmonic approximation). In reality thinks get softer with increasing temperature so g(ω) will shift down in ω as T increases.
- n<sub>BE</sub>(ω, T) is the occupation number (can think of it as an occupation probability), of a state at frequency/energy ω, and temperature T.

$$\frac{d E(\omega, T)}{dT} = \hbar \omega g(\omega) \frac{d n_{\rm BE}(\omega, T)}{dT}$$

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Derivation

$$n_{\rm BE} = \frac{1}{e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1}$$

Crazy triple chain rule

$$\frac{dn_{\rm BE}}{dT} = \frac{(-1)}{\left(\frac{\hbar\omega}{e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1}\right)^2} \times e^{\frac{\hbar\omega}{k_{\rm B}T}} \times \frac{(-1)\hbar\omega}{k_{\rm B}T^2} = \frac{\hbar\omega}{k_{\rm B}T^2} \frac{e^{\frac{\hbar\omega}{k_{\rm B}T}}}{\left(\frac{\hbar\omega}{e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1}\right)^2}$$

$$C(\omega) = \hbar\omega g(\omega) \frac{dn_{\rm BE}}{dT} = \hbar\omega \times \frac{3}{2\pi^2} \frac{\omega^2}{v_{\rm p}^2 v_{\rm g}} \times \frac{\hbar\omega}{k_{\rm B}T^2} \frac{e^{\frac{\hbar\omega}{k_{\rm B}T}}}{\left(e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1\right)^2}$$

$$C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\rm g} v_{\rm p}^2} \frac{\hbar^2}{k_{\rm B} T^2} \frac{e^{\hbar\omega/k_{\rm B} T}}{(e^{\hbar\omega/k_{\rm B} T} - 1)^2}$$

Heat capacity per volume  

$$C = \int_{0}^{\omega_{\text{max}}} C(\omega) \ d\omega$$

#### Limiting behavior

Let's look at the low  $\omega$  (and high T) behavior

$$C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\rm g} v_{\rm p}^2} \frac{\hbar^2}{k_{\rm B} T^2} \frac{e^{\hbar\omega/k_{\rm B} T}}{(e^{\hbar\omega/k_{\rm B} T} - 1)^2}$$

If I just plug in  $\omega = 0$ ,  $C(\omega) = 0/0$ . Not super helpful, what we really want is the limiting behavior.

Define: 
$$x = \frac{\hbar\omega}{k_{\rm B}T}$$
 Taylor expand  $e^x$  about  $x = 0$ .  $e^x \approx 1 + x + \cdots$   
 $C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\rm p}^2 v_{\rm g}} \frac{\hbar^2}{k_{\rm B}T^2} \frac{(1)}{(1 + x - 1)^2} = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\rm p}^2 v_{\rm g}} \frac{\hbar^2}{k_{\rm B}T^2} \left(\frac{k_{\rm B}T}{\hbar\omega}\right)^2 = k_{\rm B} \frac{3}{2\pi^2} \frac{\omega^2}{v_{\rm p}^2 v_{\rm g}}$ 
  
At high T and or low  $\omega$   
 $C(\omega) = k_{\rm B} g(\omega)$   
which means  $C \to cst$  at high T  
and  $C(\omega) \propto \omega^2$  when the dispersion is linear

#### Limiting behavior

Let's look at **low** T behavior

Convenient to convert the integral to be over the dimensionless parameter x.

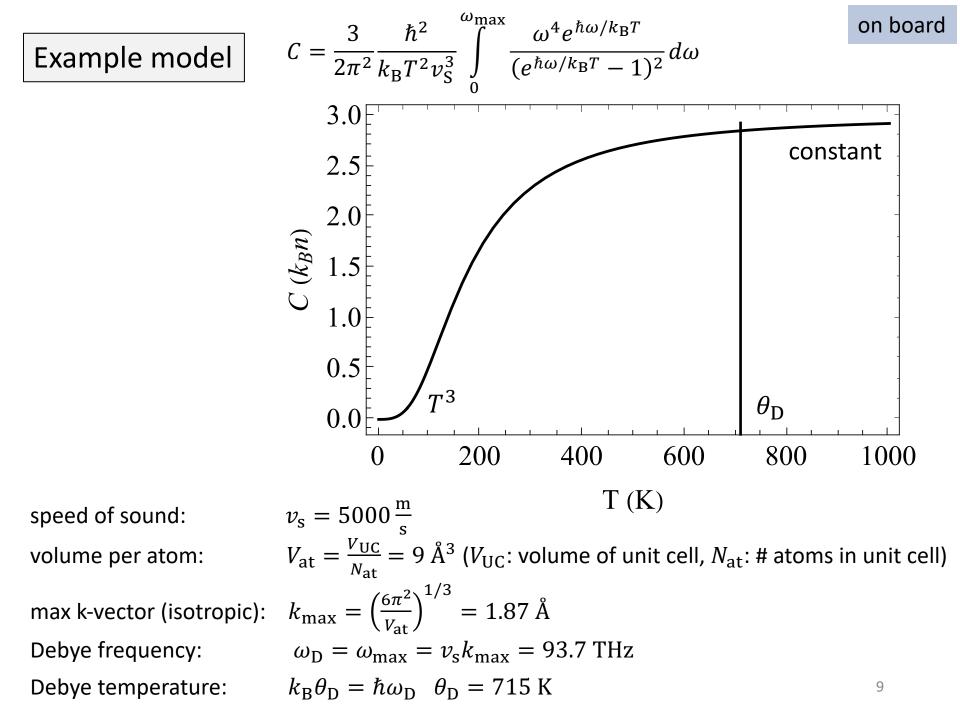
#### At low T:

1. only low frequency modes are populated,  $v_{\rm g} = v_{\rm p} = v_{\rm s}$ 

2.  $x_{\max} \to \infty$ 

$$\int_{0}^{\infty} \frac{x^4 e^x}{(e^x - 1)^2} dx = \frac{4\pi^4}{15} \qquad C = \frac{3}{2\pi^2} \frac{k_B^4 T^3}{\hbar^3} \left(\frac{4\pi^4}{v_s^3 15}\right)$$

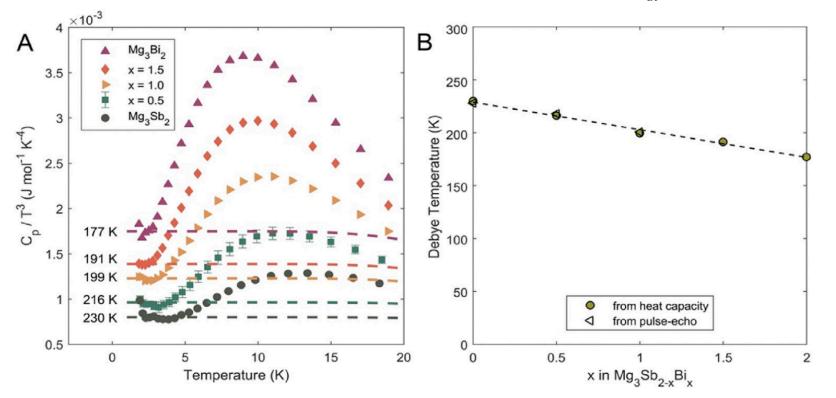
At low T where dispersion is linear:  $C \propto T^3$ 



#### Analyzing low-T data

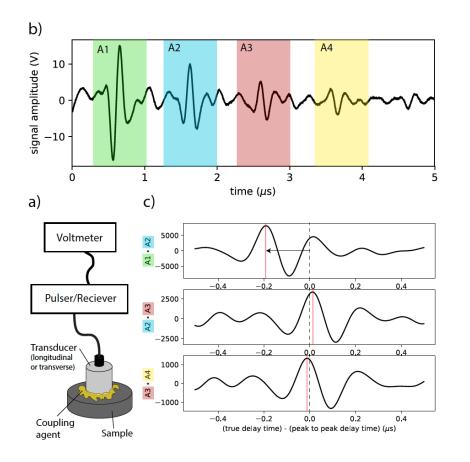
$$C = \frac{3}{2\pi^2} \frac{k_{\rm B}^4 T^3}{\hbar^3} \left( \frac{4\pi^4}{v_{\rm S}^3 15} \right)$$
  
good  
$$\frac{C}{T^3} = \left( \frac{2}{5} \right) \frac{\pi^2 k_{\rm B}^4}{\hbar^3 v_{\rm S}^3} = \beta \xrightarrow{\text{exercise}} \theta_D = \left( \frac{12\pi^4}{5} \frac{n k_{\rm B}}{\beta} \right)^{1/3}$$

 $n = \frac{1}{V_{at}}$ : number density of atoms



Agne, M. T. *et al.* Heat capacity of Mg3Sb2, Mg3Bi2 and their alloys at high temperature. *Mater. Today Phys.* **6**, DOI 10.1016/j.mtphys.2018.10.001 (2018).

# Speed of sound



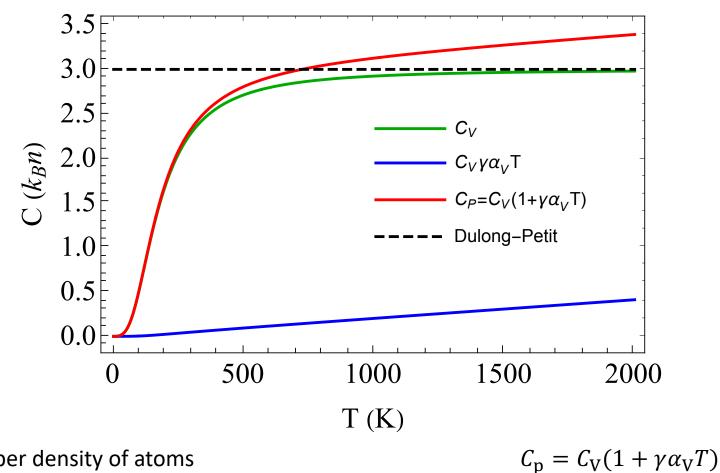
The isotropic average speed of sound, used for thermal conductivity modeling, heat capacity modeling, Debye temperature.

- When the cross-correlation is used to measure the time delay, the error is typically limit by the thickness variation of the sample. Can get better than 1% accuracy, if careful.
- If sample is a randomly oriented polycrystalline, then you are measuring the isotropic average longitudinal and transverse speeds of sound.
- Be mindful of preferred orientation of grains, especially for non-cubic materials.
- Be mindful of sample density. Samples should be very dense, >98%.

$$v_{\rm S} = \left(\frac{1}{3} \left[\frac{1}{v_{\rm L}^3} + \frac{2}{v_{\rm T}^3}\right]\right)^{-1/3}$$

Anderson, O. L. A simplified method for calculating the Debye temperature from elastic constants. J. Phys. Chem. Solids 24, 909–917 (1963).

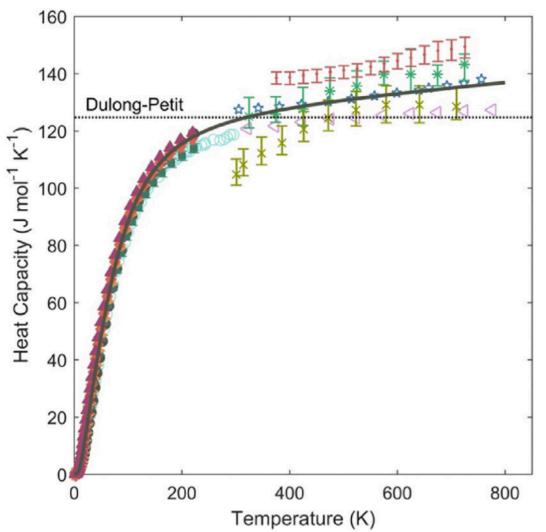
### Thermal expansion contribution



*n*: number density of atoms  $\gamma = B\alpha_V/C_V$ : Gruneisen parameter *B*: isothermal Bulk modulus  $\alpha_V$ : volumetric thermal expansion ( $\alpha_V \cong 3\alpha_L$ )

# Analyzing high-T data

- Absolute value of high temperature heat capacity is notoriously hard to measure.
- In contrast low-temperature heat capacity can be very accurate (e.g. physical property measurement system <u>PPMS</u> in heat capacity mode).
- Often times our models are better at predicting it then and one given measurement is.
- The slope of Heat Capacity vs. T at high T can be reasonably accurate.



Agne, M. T. *et al.* Heat capacity of Mg3Sb2, Mg3Bi2 and their alloys at high temperature. *Mater. Today Phys.* **6**, DOI 10.1016/j.mtphys.2018.10.001 (2018).

#### Thermal conductivity considerations

Modeling:

$$\kappa = \int C(\omega) D(\omega) d\omega$$
$$D(\omega) = \frac{1}{3} v_{g}^{2}(\omega) \tau(\omega)$$

 $C(\omega)$ : spectral heat capacity per unity volume  $D(\omega)$ : spectral diffusivity **Experiment:** 

$$\kappa = C_{\rm V} D = C_{\rm P} \rho D$$

*D*: thermal diffusivity measured by, for example LFA  $C_V$ : heat capacity per unit volume  $C_P$ : heat capacity per unit mass  $\rho$ : mass density

$$C_{\rm V} = \int C(\omega) \, d\omega$$
  

$$D = \frac{\int C(\omega)D(\omega)d\omega}{\int C(\omega)d\omega}$$
  
Heat capacity weighted  
average diffusivity  

$$\kappa = \left(\int C(\omega) \, d\omega\right) \frac{\int C(\omega)D(\omega)d\omega}{\left(\int C(\omega)d\omega\right)} = \int C(\omega)D(\omega)d\omega$$

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